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### Deposited in DRO:

14 October 2016

### Version of attached file:

Accepted Version

### Peer-review status of attached file:

Peer-reviewed

### Citation for published item:

Lechleitner, F.A. and Baldini, J.U.L. and Breitenbach, S.F.M. and Fohlmeister, J. and McIntyre, C. and Goswami, B. and Jamieson, R.A. and van der Voort, T.S. and Prufer, K. and Marwan, N. and Culleton, B.J. and Kennett, D.J. and Asmerom, Y. and Polyak, V. and Eglinton, T.I. (2016) 'Hydrological and climatological controls on radiocarbon concentrations in a tropical stalagmite.', *Geochimica et cosmochimica acta.*, 194 . pp. 233-252.

### Further information on publisher's website:

<https://doi.org/10.1016/j.gca.2016.08.039>

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# Hydrological and climatological controls on radiocarbon concentrations in a tropical stalagmite

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## Abstract

Precisely-dated stalagmites are increasingly important archives for the reconstruction of terrestrial paleoclimate at very high temporal resolution. In-depth understanding of local conditions at the cave site and of the processes driving stalagmite deposition is of paramount importance for interpreting proxy signals incorporated in stalagmite carbonate. Here we present a sub-decadally resolved dead carbon fraction (DCF) record for a stalagmite from Yok Balum Cave (southern Belize). The record is coupled to parallel stable carbon isotope

( $\delta^{13}\text{C}$ ) and U/Ca measurements, as well as radiocarbon ( $^{14}\text{C}$ ) measurements from soils overlying the cave system. Using a karst carbon cycle model we disentangle the importance of soil and karst processes on stalagmite DCF incorporation, revealing a dominant host rock dissolution control on total DCF. Covariation between DCF,  $\delta^{13}\text{C}$ , and U/Ca indicates that karst processes are a common driver of all three parameters, suggesting possible use of  $\delta^{13}\text{C}$  and trace element ratios to independently quantify DCF variability. A statistically significant multi-decadal lag of variable length exists between DCF and reconstructed solar activity, suggesting that solar activity influenced regional precipitation in Mesoamerica over the past 1500 years, but that the relationship was non-static. Although the precise nature of the observed lag is unclear, solar-induced changes in North Atlantic oceanic and atmospheric dynamics may play a role.

## **1. Introduction**

Stalagmites are critical archives for the reconstruction of terrestrial paleoclimate. They are dateable with exceptional precision, and provide high-resolution time series data that reflect past climatic and environmental conditions (e.g., Ridley et al., 2015a; Vaks et al., 2013). However, because local conditions that influence proxy signals can vary between cave sites, careful interpretation of stalagmite paleoclimate records is necessary. A robust interpretation of stalagmite paleoclimate proxies therefore requires detailed knowledge of surface and cave conditions, including cave monitoring studies (Breitenbach et al., 2015), and assessments of hydrological and carbon cycle processes within the karst system (Frisia et al., 2011; Noronha et al., 2015; Rudzka-Phillips et al., 2013).

66 Combined analyses of stable carbon isotopes and  $^{14}\text{C}$  in stalagmite carbonate can  
67 be particularly informative because the two proxies reflect carbon inputs from  
68 different surface environment sources (atmosphere, soil and vegetation), and  
69 from the host rock (Genty et al., 2001; Hendy, 1971; Oster et al., 2010). Meteoric  
70 water encounters high  $\text{CO}_2$  levels in the soil, epikarst, and bedrock atmosphere  
71 (Baldini, 2010; Breecker et al., 2012; Noronha et al., 2015). Due to the biological  
72 nature of the processes involved in the production of soil  $\text{CO}_2$  (microbial  
73 decomposition of soil organic matter (SOM) and root respiration), the  $\delta^{13}\text{C}$  is  
74 strongly depleted (around  $-26\text{‰}$  for areas dominated by  $\text{C}_3$ -type plants),  
75 whereas  $^{14}\text{C}$  is often slightly to moderately depleted compared to the  
76 contemporaneous atmosphere through the decomposition of older residual SOM  
77 (Suppl. Fig. 1) (Genty and Massault, 1999). Dissolution of the ancient (i.e.,  $^{14}\text{C}$ -  
78 free) carbonate host rock by the acidic aqueous solution results in higher  $\delta^{13}\text{C}$   
79 values but a further reduction in  $^{14}\text{C}$  contents in the water solution (Suppl. Fig. 1)  
80 (Genty et al., 2001). Carbonate speleothems form when dripwater saturated with  
81 respect to  $\text{CaCO}_3$  enters a cave, where  $\text{CO}_2$  levels are generally much lower than  
82 in the dripwater solution (McDermott, 2004).  $\text{CO}_2$  degassing leads to  
83 supersaturation in the solution with respect to  $\text{CaCO}_3$  and subsequent carbonate  
84 precipitation. Rapid degassing, for example in well-ventilated caves or under  
85 slow drip rates, promotes kinetic isotopic fractionation effects, leading to  
86 substantially higher  $\delta^{13}\text{C}$  values (Breitenbach et al., 2015; Frisia et al., 2011).  
87 Early studies attempting to date groundwater using  $^{14}\text{C}$  concluded that the  
88 composite origin of groundwater carbon leads to large age offsets compared to  
89 the contemporaneous atmosphere (Fontes and Garnier, 1979; Wigley, 1975),  
90 which is then transferred to stalagmite carbonate. The difference between the

91 stalagmite and the contemporaneous atmosphere  $^{14}\text{C}$  content at the time of  
92 carbonate deposition is called the 'dead carbon fraction' (DCF), and can be highly  
93 variable depending on karst and soil conditions, such as the thickness of bedrock  
94 overlying the cave and SOM age spectrum (Genty et al., 2001; Griffiths et al.,  
95 2012; Noronha et al., 2014; Rudzka et al., 2011). Detailed understanding of  
96 carbon cycle controls is therefore paramount for understanding specific karst  
97 systems and for the correct interpretation of stalagmite proxy records.

98 Well-dated stalagmite  $^{14}\text{C}$  time series have extended the IntCal calibration curve,  
99 taking into account DCF as a constant offset between stalagmite  $^{14}\text{C}$   
100 measurements and IntCal (Hoffmann et al., 2010; Southon et al., 2012). These  
101 studies led to significant improvements in our ability to date natural and  
102 archaeological samples in the absence of direct atmospheric  $^{14}\text{C}$  records such as  
103 tree rings (i.e., beyond 13.9 kyr BP) (Reimer et al., 2013). However, DCF  
104 variations beyond the tree-ring based interval of the calibration curve are  
105 difficult to account for and to distinguish from variations in atmospheric  $^{14}\text{C}$   
106 activity, requiring a method independent from the calibration curve for the  
107 detection of DCF variations in stalagmites. Although DCF may be relatively  
108 constant in a cave environment over long periods of time (e.g., in stalagmite H-82  
109 from Hulu Cave; Southon et al., 2012), significant short-term variations can occur  
110 (Griffiths et al., 2012; Noronha et al., 2014), especially during climatic extremes  
111 (e.g., the last deglaciation; Oster et al., 2010; Rudzka et al., 2011). Understanding  
112 the factors driving DCF variations would not only be important for calibration  
113 purposes, but might also open the door to  $^{14}\text{C}$  dating of stalagmites using  
114 conventional calibration approaches.

Here we present a sub-decadally resolved stalagmite  $^{14}\text{C}$  record from the tropical Yok Balum Cave, Belize. The exceptional resolution and chronological precision of our  $^{14}\text{C}$  record allows direct comparison to atmospheric  $^{14}\text{C}$  activity over the past 1500 years, and provides valuable insights into how hydrology and the karst pathways respond to climatic changes at the site. We use  $\delta^{13}\text{C}$  and U/Ca to infer the importance of kinetic fractionation and prior calcite precipitation (PCP) and/or prior aragonite precipitation (PAP) occurring at the site. Carbon cycle modeling and the analysis of soil samples from above the cave help disentangle the main processes influencing  $^{14}\text{C}$  and  $\delta^{13}\text{C}$  at our site and strengthen the proxy interpretation. We compare our high-resolution  $^{14}\text{C}$  record to atmospheric  $^{14}\text{C}$  from IntCal13 (Reimer et al., 2013) and solar activity proxies to detect similarities and infer driving mechanisms.

## **2. Cave setting and climate**

Yok Balum Cave is located in southern Belize in the district of Toledo (16°12'30.780 N, 89°4'24.420 W, 366 m above sea level) (Fig. 1). The cave developed in a steep and remote hill in a SW-NE trending karst ridge composed of limestone of Cretaceous age of the Campur Formation (Kennett et al., 2012; Miller, 1996). The vegetation above the cave consists of dense subtropical forest, composed primarily of C3 plants. Soil thickness above Yok Balum Cave varies considerably; it is generally very thin (< 30 cm) but occasionally forms deeper (up to 60 cm) pockets in the strongly karstified limestone. The soil is a leptosol (WRB, 2006) and has poorly developed horizons. Due to the generally inaccessible location of the hilltop above Yok Balum Cave, it is unlikely that the vegetation and cave hydrology was ever disturbed by farming activities in the

past, although the area has been populated for millennia (Kennett et al., 2012; Walsh et al., 2014).

Yok Balum Cave consists of a single main trunk conduit overlain by ~ 14 m of karstified bedrock with one entrance at each end at different elevations, resulting in constant airflow and a dynamic diurnal and seasonal ventilation regime (Ridley et al., 2015a, 2015b) (Fig. 1). Detailed cave microclimate monitoring, including logging of temperature, cave air CO<sub>2</sub>, radon, and drip rates, has been carried out since 2011 (Kennett et al., 2012; Ridley et al., 2015b). The cave has a nearly constant temperature of  $22.9 \pm 0.5^{\circ}\text{C}$  (Ridley et al., 2015b) that closely reflects the outside mean annual air temperature. Belize is located at the northernmost extent of the present-day boreal summer Intertropical Convergence Zone (ITCZ), whose annual migration dominates local climate (Ridley et al., 2015a) (Fig. 1). Precipitation is heavily biased towards the boreal summer months, when 400-700 mm of monthly rainfall can be registered, whereas winters are generally very dry ( $< 70$  mm/month; Poveda et al., 2006).

### **3. Materials and methods**

#### **3.1. Stalagmite YOK-I**

Stalagmite YOK-I was collected in 2006 and is 606.9 mm long. The upper 415 mm are entirely composed of aragonite and were analyzed previously for high resolution stable isotopes of oxygen ( $\delta^{18}\text{O}$ ) and  $\delta^{13}\text{C}$  (Kennett et al., 2012) (Fig. 2). YOK-I was actively growing at the time of collection, and detailed U-Th measurements indicate that the aragonitic section spans the last 2000 years (Kennett et al., 2012). In this study, the top 285.5 mm of YOK-I were resampled for  $^{14}\text{C}$ ,  $\delta^{13}\text{C}$ , and U/Ca.

166

### 167 **3.2. Stalagmite <sup>14</sup>C measurements**

168 Samples for high-precision graphite <sup>14</sup>C analysis were milled continuously along  
169 the growth axis, following the previous stable isotope sampling transect, using a  
170 semi-automated high-precision drill (Sherline 5400 Deluxe) at ETH Zürich. The  
171 resultant transect produced 198 high-precision <sup>14</sup>C measurements, taken at a  
172 resolution between 0.5 – 3.3 mm. Contamination from sample and equipment  
173 handling was minimized by cleaning all surfaces with methanol and drying using  
174 compressed air between each sample. Additionally, the top 0.1 mm of stalagmite  
175 surface was discarded after milling. Graphitization and <sup>14</sup>C analysis were  
176 performed at the Laboratory for Ion Beam Physics (LIP) at ETH Zürich. 8-12 mg  
177 aliquots of carbonate powder were graphitized using an automatic  
178 graphitization system fitted with a carbonate handling system (CHS-AGE,  
179 Ionplus) and <sup>14</sup>C content was measured with an accelerator mass spectrometer  
180 (MICADAS, Ionplus). Oxalic acid II (NIST SRM 4990C) was used as the  
181 normalizing standard and was measured to a precision better than 2‰. IAEA-C1  
182 was used as blank while IAEA-C2 and a modern coral standard were used as  
183 secondary standards. A <sup>14</sup>C-free stalagmite sample was used as a processing  
184 control.

185

### 186 **3.3. Stable isotope and trace element analysis**

187 YOK-I was previously sampled at 100 µm resolution for δ<sup>13</sup>C and δ<sup>18</sup>O analysis,  
188 published in Kennett et al (2012). To avoid any depth bias during the re-  
189 sampling for the current study, stable isotope measurements were performed on  
190 aliquots from some of the same powders. This was especially important because



the age model based on the stable isotopes was applied to this study. Samples were analyzed for  $\delta^{18}\text{O}$  and  $\delta^{13}\text{C}$  on a Thermo Delta V Plus mass spectrometer coupled with a ThermoFinnigan GasBench II carbonate preparation device at the Geological Institute, ETH Zürich, following the procedure described in Breitenbach and Bernasconi (2011).

U/Ca ratios were measured on aliquots of the same powders used for  $^{14}\text{C}$  and  $\delta^{13}\text{C}$ . The powders were dissolved in 1% Nitric Acid (PWR 67% Nitric Acid Ultrapure Normatom for trace element analysis, diluted with ultrapure water) and measured using a Thermo Scientific X-Series II inductively-coupled plasma mass spectrometer (ICP-MS) at Durham University. Multi-elemental Romil standards and blanks were run throughout the sequence to allow precise quantification and correction for machine drift. Analytical precision for U was <5% RSD for individual measurements, and detection limits were generally <1 ppt. Ca measurement precision was generally <2% RSD, with all analyses well above detection limits of  $\sim$ <0.1ppb.

### **3.4. Soil samples**

A ca. 60 cm deep soil profile, extending to the top of the bedrock, was collected in June 2013. Because of the extreme karstification of the bedrock, soil thickness is very variable above the cave, and a deeper pocket was chosen to capture the maximum extent of the soil. The profile was sampled at 4-5 cm per sampled depth for a total of 13 samples. All samples were stored in dark and cool conditions whenever possible, and freeze-dried upon arrival to the laboratory. Smaller aliquots of the soil samples were homogenized and larger plant fragments particles (> 5 mm) were removed manually.

216 The amount of organic carbon and  $^{14}\text{C}$  content in the soil profile was determined  
217 at LIP, ETH Zürich. To remove carbonates prior to analysis, aliquots of  
218 homogenized soil samples were transferred to silver capsules and fumigated  
219 over three days at 60°C using 37% HCl (puriss. p.a. grade, Sigma Aldrich)  
220 (Komada et al., 2008) and neutralized for 24 hrs over solid NaOH. Samples were  
221 then wrapped in a second tin capsule and pressed. The % organic carbon was  
222 determined using an elemental analyzer (Vario MICRO cube, Elementar)  
223 calibrated using atropine as the standard (Säntis, product SA990746B).  $^{14}\text{C}$   
224 content was determined on a second aliquot of carbonate-free soil containing 1  
225 mg carbon using an automated graphitization system and an accelerator mass  
226 spectrometer (AGE-MICADAS, Ionplus). Oxalic acid II (NIST SRM 4990C) was the  
227 normalizing standard measured to 4‰ precision. Ancient anthracite coal was  
228 used as the blank and processing control and IAEA-C7 and -C8 were used as  
229 secondary standards. Samples were corrected for constant contamination by  
230 extraneous carbon using the anthracite processing control and secondary  
231 standards.

232 Water extractable organic carbon (WEOC) of the soil samples was characterized  
233 to infer the nature of SOM transported through the karst.  $^{14}\text{C}$  content of WEOC  
234 was determined by extracting 5 g of soil with 20 ml of 0.5 wt% NaCl (in ultrapure  
235 water) in pre-combusted glass centrifuge tubes (similar to Hagedorn et al.,  
236 2004). The tubes were centrifuged three times for 15 min, and the solution was  
237 re-homogenized using a vortex mixer in between. The supernatant was decanted  
238 using combusted glass pipettes, filtered through a column containing a small  
239 amount of pre-combusted glass fibre to remove solid particles, and freeze-dried  
240 using a Christ Alpha 1-2 LD plus freeze-dryer equipped with an oil-free pump to

prevent contamination. The extracts were then transferred to pre-combusted 12 ml borosilicate Exetainer vials (Labco) using 5 ml of ultrapure water at pH 2. The  $^{14}\text{C}$  content was measured following the method described in Lang et al. (2016) using wet chemical oxidation and accelerator mass spectrometry using a Gas Ion Source (GIS) interface (Ionplus).

### 3.5. Carbon isotope models

DCF in stalagmite YOK-I ( $\text{DCF}_{\text{YOK-I}}$ ) reflects various sources, mainly soil and vegetation, carbonate host rock, and fractionation effects (Griffiths et al., 2012). In order to separate and infer the relative importance of each contributing source to  $\text{DCF}_{\text{YOK-I}}$ , a modified version of a soil-karst carbon isotope model, described in Fohlmeister et al. (2011) and Griffiths et al. (2012), was applied to the dataset (Suppl. Fig. 2). Briefly, the model first calculates the SOM spectrum that best fits the measured stalagmite bomb spike. Three SOM pools with different mean ages and turnover times were calculated, and optimized to find the best fit with the measured stalagmite bomb spike via a Monte Carlo approach (30,000 runs). The SOM spectrum was applied to the entire dataset to reveal the  $^{14}\text{C}$  content of the soil gas. This assumes that vegetation and soil composition have remained constant over the period of stalagmite growth. Fractionation effects between  $\text{CO}_2$  and  $\text{HCO}_3^-$  when entering the groundwater DIC solution are taken into account using the fractionation factor for  $^{14}\text{C}$   $^{14}\epsilon = 2 \times ^{13}\epsilon/10$  (Southon, 2011), resulting in approximately +1.8 fraction modern ( $\text{F}^{14}\text{C}$ ; Reimer et al., 2004) at 25°C. The remaining DCF signal is divided into host rock dissolution and in-cave kinetic fractionation effects. In-cave kinetic fractionation ( $\Delta\delta^{13}\text{C}$ ), including the effects of PCP/PAP, is calculated as the difference between

stalagmite  $\delta^{13}\text{C}$  and  $\delta^{13}\text{C}$  estimated for the drip water solution after carbonate dissolution, when water is saturated with respect to  $\text{Ca}^{2+}$  (Griffiths et al., 2012). Dripwater  $\delta^{13}\text{C}$  can be calculated iteratively, by considering the host rock  $\delta^{13}\text{C}$  and the soil-water DIC  $\delta^{13}\text{C}$  (in our case 0‰ and -17‰, respectively). Using the DCF value at that point in time permits calculation of the relative contribution of the host rock and DIC to the total dripwater  $\delta^{13}\text{C}$  (as described in Griffiths et al, 2012). Kinetic fractionation effects on  $^{14}\text{C}$  are readily quantifiable, because a 1‰ change in  $\delta^{13}\text{C}$  equals a shift of ca. 0.2  $\text{F}^{14}\text{C}$  in  $^{14}\text{C}$  (Southon, 2011). After removing the effects of vegetation/SOM and kinetic fractionation, the residual DCF is attributed to host rock dissolution processes.

## **4. Results**

### **4.1. YOK-I $^{14}\text{C}$ record**

The YOK-I  $^{14}\text{C}$  record extends from  $\sim -54$  back to 1400 years BP (i.e., 2004 to 555 C.E.), based on the U/Th age model constructed by Kennett et al. (2012) (Table 1, Fig. 2). A gap is present between 1341 and 1400 C.E., due to sampling difficulties at the transition between two slabs of stalagmite YOK-I. The mean temporal resolution is 5 years, and the maximum resolution is 0.7 years. A general decay trend is visible between 555 and 1950 C.E., with superimposed deviations in the range of  $\pm 0.2 \text{ F}^{14}\text{C}$ . The modern part of the  $^{14}\text{C}$  record (1950-present, top 9.3 mm) shows a clear imprint of bomb carbon, with maximum values of 1.14  $\text{F}^{14}\text{C}$  (at 1990 C.E.) (Fig. 2B).

Conversion of  $^{14}\text{C}$  activity to DCF reveals significant variability over the entire interval studied (Table 1, Fig. 3A). Errors in  $\text{DCF}_{\text{YOK-I}}$  are between 0.23 and 0.67%, and were calculated using error propagation following Noronha et al.

(2014).  $DCF_{YOK-I}$  values range between 9.04 and 16.7% (mean: 12.9%). The lowest DCF values occur during the period ca. 700-1100 C.E., concurrent with the most enriched  $\delta^{13}C$  values (Fig. 3C).

#### **4.2. Stable isotopes and U/Ca**

The new  $\delta^{13}C$  record measured on aliquots of the samples used for  $^{14}C$  and U/Ca analyses, is of a lower resolution but shows excellent agreement with the previous high-resolution profile published in Kennett et al. (2012), confirming that no spatial error occurred during the resampling (Table 1, Fig. 3C). Several pronounced positive excursions in  $\delta^{13}C$  are apparent throughout the record, e.g., at ca. 1780, 1500, 940, 620, and most notably, between 1040-1100 C.E.

184 aliquots of powders drilled for  $^{14}C$  analysis were also used for U/Ca measurements. Values (expressed as U/Ca in ppm/ppm  $\times$  1000) vary from 0.00068 to 0.02952, with a pronounced minimum during the period 1040-1100 C.E. and highest values at the beginning of the record (550-700 C.E.) (Table 1, Fig. 3B). A large gap in U/Ca measurements exists between  $\sim$  1250-1600 C.E., due to the transition between two stalagmite slabs (as in the  $^{14}C$  record), as well as lack of availability of sufficient sample powder for analysis. The early part of the record (550-1180 C.E.) is generally characterized by pronounced variability in U/Ca with several rapid (sub-decadal) large excursions synchronous with shifts in  $\delta^{13}C$ , whereas the more recent part (1600-1950 C.E.) shows much more uniform values.

#### **4.3. Soil samples**

Soil organic carbon (SOC) content was measured twice with similar results (Table 2, Fig. 4, series A and B). The highest values are found in the top sample (~20% organic carbon), mainly composed of plant litter in the organic horizon, followed by a steady decrease towards the bottom of the profile, with the lowest sample (at ~60 cm depth) only containing ~2% organic carbon.

$F^{14}C$  values from the bulk SOC are generally quite high (0.85 to 1.1  $F^{14}C$ ), with systematically decreasing values towards the bottom of the profile (Table 2, Fig. 4). The presence of bomb carbon is suggested at the top of the profile, where the highest values are found between 5-15 cm below the surface, whereas in the topmost sample,  $F^{14}C$  is slightly lower. The WEOC  $F^{14}C$  shows a similar pattern as the bulk soil, with a steady decrease in  $F^{14}C$  from the top to the bottom of the profile (0.93 to 1.09  $F^{14}C$ ). There is a bifurcation in WEOC and bulk SOC  $F^{14}C$  values with increasing depth, with the WEOC fraction decreasing less rapidly and implying younger carbon than in the bulk SOC (Table 2, Fig. 4).

#### **4.4. Karst carbon isotope modeling**

The model with the best fit to the bomb spike data from YOK-I (Fig. 5A) produces a SOM spectrum with the following parameters:

$y_1 = 6$  years;  $c_1 = 34\%$

$y_2 = 37$  years;  $c_2 = 62\%$

$y_3 = 580$  years;  $c_3 = 4\%$

where  $y_i$  denotes the mean age of the SOM pools and  $c_i$  the relative contribution of the SOM pools to the respired soil gas  $CO_2$ . Applying this spectrum to the entire record shows that most of the atmospheric variation is expressed in the soil gas, due to the young SOM spectrum (Fig. 5B). Nevertheless, soil gas  $^{14}C$

activity is  $\sim 0.01$   $F^{14}C$  lower than the contemporaneous atmospheric  $^{14}C$  activity, and lagging the latter by  $\sim 15$  years (Fig. 5B), due to the integrating nature of SOM. A slight enrichment occurs due to fractionation effects between  $CO_2$  and DIC in the soil. The average contribution from vegetation and SOM to  $DCF_{YOK-I}$  is  $0.015 F^{14}C$ , whereas the average enrichment from in-cave fractionation is  $-0.027 F^{14}C$ . The host rock contribution is therefore dominant, amounting to  $0.139 F^{14}C$  on average (Fig. 5C).

## **5. Discussion**

### **5.1. Sources of carbon in stalagmite YOK-I**

We disentangle the influence of soil and karst processes on stalagmite carbon isotopes by combining high-precision isotope measurements on stalagmites, bulk SOC and soil WEOC, and karst carbon isotope modeling.

The trend towards lower  $^{14}C$  activities in the soil profile (Fig. 4) reflects general ageing of the SOM related to gradual soil buildup, and the slow downward cycling of dissolved organic matter (DOM), as described in a conceptual model by Kaiser and Kalbitz (2012). In this model, temporary storage of DOM through sorption mechanisms and microbial degradation result in an increasing trend in SOM  $^{14}C$  ages with depth. The WEOC represents the most labile pool of SOM that is readily dissolved in water (Hagedorn et al., 2004) and reflects the same trend as the bulk soil samples, but with a less pronounced decrease in  $^{14}C$  content. This is likely related to the preferential extraction of smaller, and thus more labile, compounds from the soil, including those from living microbial biomass (Hagedorn et al., 2004; Jones and Willett, 2006).

The analysis of bulk soil and WEOC samples shows that the SOM spectrum from the soil above Yok Balum is quite young, and that the DOM leached from the soil matrix (WEOC) echoes this trend. Backward modeling of SOM from the bomb spike in YOK-I corroborates a very young SOM contribution to the karst system (96% <50 years old) (Fig. 5). Although the model assumes constant vegetation type and density above Yok Balum Cave over the past 1500 years, vegetation shifts may have occurred because of severe droughts recorded between 700-1100 C.E. (Kennett et al., 2012). Less dense vegetation and reduced soil microbial activity during dry periods or under sustained deforestation would result in older apparent ages of the SOM and lead to stronger smoothing of the atmospheric  $^{14}\text{C}$  signal delivered to the cave and increased stalagmite DCF (Fohlmeister et al., 2011a). However, this signal would be opposite than that observed in  $\text{DCF}_{\text{YOK-I}}$  during the 700-1100 C.E. period, where DCF is at its minimum. We attribute this to the minor influence of SOM to  $\text{DCF}_{\text{YOK-I}}$  (Fig. 5C), and therefore we conclude that large changes in DCF cannot originate from SOM variability.

Young and fast cycling soils are often observed at tropical sites (Trumbore, 1993), where high temperature and humidity promote biological activity and consequently result in high SOM turnover rates (Davidson and Janssens, 2006). On the other hand, studies from (sub-)tropical karst settings have suggested that a substantial contribution from pools of pre-aged SOM must influence the carbon cycle at these locations: at Liang Luar Cave, on the Indonesian island of Flores, the modeled SOM was dominated by a multi-centennial carbon pool (Griffiths et al., 2012). A very old SOM spectrum was also observed in a recent study on soils from above Heshang Cave, China (Noronha et al., 2015). It is likely that



differences in local conditions, soil depth, and microbial activity, and the magnitude of pre-aged organic carbon reservoirs in the deep vadose zone, are responsible for the contrasting characteristics of the Yok Balum Cave speleothem.

The overall modeled contribution of SOM to the  $DCF_{YOK-I}$  is found to be small (max. 2.5%), and the largest contributions to  $DCF_{YOK-I}$  appear to come from carbonate dissolution in the karst and from changes in karst hydrology (Fig. 6C). Measured  $DCF_{YOK-I}$  shows substantial and rapid transitions of up to 4%, with lower DCF values correlating with less negative  $\delta^{13}C$  and  $\delta^{18}O$  values and vice-versa (Fig. 3). This suggests lower/higher  $DCF_{YOK-I}$  occurred during drier/wetter conditions, corroborating studies where stalagmite DCF was observed to co-vary with other hydroclimate proxies (Griffiths et al., 2012; Noronha et al., 2014). The hydrological imprint on DCF appears to be related to shifts between the open and closed end-members of the karst system (Hendy, 1971). More open system conditions prevail during periods of lower recharge, i.e. drier periods. This promotes lower DCF values as the karst aqueous solution constantly re-equilibrates with the soil  $CO_2$  reservoir through air-filled voids and pores, resulting in higher water (and stalagmite)  $^{14}C$  activities. Conversely, during wetter periods, the karst system is more often waterlogged and the aqueous solution becomes isolated from the contemporaneous soil  $CO_2$  reservoir (closed system), resulting in much higher amounts of dead carbon from carbonate dissolution being added to the solution (Fohlmeister et al., 2011b).

The importance of kinetic fractionation and PCP/PAP with respect to carbon isotopic signatures in YOK-I are investigated using both  $\delta^{13}C$  and U/Ca, coupled to modeling. We consider both processes; although YOK-I is aragonitic, PCP

could occur in the karst overlying the cave, increasing the Mg/Ca ratio in the aqueous solution, and consequently resulting in aragonite precipitation in the cave (Wassenburg et al., 2012). U sourced from the overlying soil and the host rock itself can be modulated by PCP/PAP (Johnson et al., 2006). Because U is incorporated in the carbonate lattice, PAP should effectively scavenge U from the drip water solution, resulting in lower stalagmite U contents during drier periods (Jamieson et al., *in press*; Wassenburg et al., *in press*).  $\delta^{13}\text{C}$  is strongly altered by kinetic in-cave fractionation and PCP/PAP, as forced degassing by low  $\text{CO}_2$  partial pressure enriches the solution in  $^{13}\text{C}$  (Frisia et al., 2011; Hendy, 1971). Modeling of kinetic fractionation effects between DIC and  $\text{CaCO}_3$  (both in-cave fractionation and PCP/PAP) in stalagmite YOK-I shows that most of the variation in  $\delta^{13}\text{C}$  is attributable to this process, whereas the soil and carbonate host rock signatures are only responsible for the overall range in  $\delta^{13}\text{C}$  (Fig. 6). Despite the fact that mass-dependent fractionation with respect to  $^{12}\text{C}$  is about twice as strong for  $^{14}\text{C}$  than  $^{13}\text{C}$ , fractionation effects are generally not as strongly expressed in  $^{14}\text{C}$  as in  $\delta^{13}\text{C}$ , due to the difference in unit of the two parameters (‰ in  $^{14}\text{C}$  vs. ‰ in  $\delta^{13}\text{C}$ ) (Fohlmeister et al., 2011b; Southon, 2011). Most of the variability in  $\delta^{13}\text{C}$  attributed to fractionation by the karst model is also present in the U/Ca record (Fig. 6B). Several large and rapid positive excursions are found both in  $\delta^{13}\text{C}$  and U/Ca, most notably between 1040 and 1100 C.E., and all coincide with periods of increased in-cave kinetic fractionation as calculated with  $\Delta\delta^{13}\text{C}$ . U/Ca ratios in YOK-I therefore are interpreted to reflect local hydrological conditions and the amount of PAP occurring at the site, providing additional evidence for kinetic fractionation as the main driver of  $\delta^{13}\text{C}$  in this stalagmite.

Previous studies have highlighted the potential of hydrological proxies for detecting past stalagmite DCF shifts: Rudzka et al. (2011) showed that shifts in DCF during the last deglaciation were matched by synchronous shifts in  $\delta^{13}\text{C}$ , implying a common forcing mechanism on the two proxies (e.g., effective infiltration or changes in mean SOM age). Another study combined DCF and Mg/Ca data measured on a tropical stalagmite and highlighted the importance of host rock dissolution processes for stalagmite DCF (Griffiths et al., 2012).

In YOK-I, both  $\delta^{13}\text{C}$  and U/Ca values show remarkable similarities ( $r = -0.83$ ,  $p < 0.001$ ), suggesting a strong imprint of PCP/PAP and in-cave kinetic fractionation on both proxies. Comparison with  $\text{DCF}_{\text{YOK-I}}$  reveals a significant correlation with respect to U/Ca ( $r = 0.49$ ,  $p < 0.001$ ) and  $\delta^{13}\text{C}$  ( $r = -0.5$ ,  $p < 0.001$ ), suggesting a common forcing on all three proxies (Fig. 7). Since kinetic fractionation is not a strong component of  $\text{DCF}_{\text{YOK-I}}$  (Fig. 5), another mechanism driven by the same forcing that controls U/Ca and  $\delta^{13}\text{C}$  must exist. The modeling results confirm that, similar to previous studies, the dominant control on  $\text{DCF}_{\text{YOK-I}}$  is the dissolution of host rock carbonate, driven by open vs. closed system conditions. All three processes (host rock dissolution, kinetic fractionation and PCP/PAP) are sensitive to effective infiltration within the karst, and thus ultimately driven by climatic conditions. Increasing aridity leads to more open-system conditions and enhanced PAP and kinetic fractionation, resulting in strong covariance between DCF, U/Ca and  $\delta^{13}\text{C}$  (Fig. 7). This relationship highlights the potential usefulness of combined  $\delta^{13}\text{C}$ , trace element and  $^{14}\text{C}$  records to infer past DCF variability. It may also be possible to detect changing infiltration even when DCF cannot be readily calculated, i.e., during time intervals beyond the tree-ring based interval of the  $^{14}\text{C}$  calibration curve or for  $^{14}\text{C}$  dating applications. U/Ca

ratios are increasingly recognized as sensitive tracers for PAP in aragonitic stalagmites (Jamieson et al., *in press*), and other trace elements have successfully been used in calcitic samples (e.g., Mg/Ca, Griffiths et al., 2012).

Detailed analysis of the sources of carbon in YOK-I reveals a strong dependency on both hydroclimate and the amount of effective infiltration into the karst system. DCF,  $\delta^{13}\text{C}$  and U/Ca all show a trend towards drier conditions during the period 700-1100 C.E., a time interval previously described in conjunction with the disintegration of Classic Maya political systems (Douglas et al., 2015; Haug et al., 2003; Hodell et al., 1995; Kennett et al., 2012). Whereas  $\delta^{18}\text{O}$  reflects the amount of precipitation, moisture source and storm path length,  $\delta^{13}\text{C}$  is a useful local indicator of effective infiltration into the karst (Ridley et al., 2015a). All factors driving  $\delta^{13}\text{C}$  result in its enrichment during dry periods: reduced vegetation density and soil microbial activity result in higher  $\delta^{13}\text{C}$  values of the soil water; more open-system conditions in the karst promote PCP/PAP and kinetic fractionation, progressively enriching  $\delta^{13}\text{C}$  in the aqueous solution. Therefore, although the kinetic nature of the processes acting on  $\delta^{13}\text{C}$  prevent quantification of the hydrological deficit,  $\delta^{13}\text{C}$  in YOK-I is a sensitive recorder of infiltration dynamics.

## 5.2. 'Bomb' radiocarbon signals YOK-I

The young SOM contribution to the drip water at Yok Balum Cave is reflected in the pronounced bomb spike in stalagmite YOK-I, which reaches its peak at 1.14  $\text{F}^{14}\text{C}$ , with an overall spike of 0.27  $\text{F}^{14}\text{C}$  (Fig. 2B). Comparing this value to the maximum  $\text{F}^{14}\text{C}$  in the atmospheric Northern Hemisphere zone 2 record (1.98  $\text{F}^{14}\text{C}$  in 1963; Hua et al., 2013) confirms that YOK-I is a highly responsive

stalagmite in terms of carbon transfer, with a damping ratio,  $D$ , of 66.1%.  $D$  is calculated as the difference between the highest and lowest bomb- $^{14}\text{C}$  value in the stalagmite, compared to the atmospheric value. In comparison to an extensive study of a number of stalagmites by Rudzka-Phillips et al. (2013), YOK-I shows one of the least dampened bomb spikes. The rapid increase in  $F^{14}\text{C}$ , synchronous with the beginning of the bomb spike rise, also highlights the rapid fluid transfer in the karst at Yok Balum Cave. These features could be related to the much higher sampling resolution in YOK-I compared to other studies; however, the strong similarity between the bomb spike recorded in YOK-I and YOK-G, another stalagmite from the same cave (Ridley et al., 2015a), suggests that the amplitude of the perturbation is real. The YOK-I bomb spike does not show a pronounced maximum, but rather a rapid increase in  $^{14}\text{C}$  activity until ca. 1970 C.E., followed by a plateau, until decrease slowly starts after ca. 1990 C.E., very similar to YOK-G (Fig. 2B). It is worth noting that the measured drip rate for stalagmite YOK-G was 30 times higher than for YOK-I, likely attributable to different hydrological pathways in the karst overlying the cave (Ridley et al., 2015a). This corroborates the notion that processes related to the turnover of soil organic matter are responsible for the modulation of the bomb spike in stalagmites (Genty and Massault, 1999; Rudzka-Phillips et al., 2013), rather than changes in karst hydrology. The two stalagmite bomb spikes from Yok Balum Cave and the resultant modeled SOM spectra support the results from the analysis of soil and WEOC samples, indicating only minor contributions of old recalcitrant carbon from the soil to the karst system. Compared again with the study by Rudzka-Phillips et al. (2013), the stalagmites from Yok Balum Cave show similar behavior to the samples from arid and warm sites, with sparse

vegetation and thin soils. Although southern Belize is not characterized by year-round aridity, the boreal winter months are very dry, and infiltration in the karst is significantly reduced (Ridley et al., 2015b). Together with the low carbon storage potential of the soils overlying Yok Balum Cave, this may explain the apparent similarity to the arid sites described in Rudzka-Phillips et al. (2013).

### **5.3. Lagged solar influence on DCF**

Similarities are apparent when comparing  $DCF_{YOK-I}$  to proxies for solar activity (which modulates the production rate of atmospheric  $^{14}C$ ; Abreu et al., 2013), such as the total solar irradiance (dTSI) record by Steinhilber et al. (2009) (Fig. 8). A lag-correlation analysis was performed between YOK-I and the Steinhilber dTSI record. The YOK-I DCF and  $\delta^{13}C$  records were first estimated on the same (uniformly sampled) time scale as that of the Steinhilber dTSI using a Bayesian proxy estimation approach presented in Goswami et al. (2014). All records were normalized to mean zero and unit standard deviation, following which a millennial trend was removed and the resulting residuals were smoothed with a Gaussian kernel of 5 years width. Pearson's cross correlation was then estimated between the resulting smoothed residuals at different lags by shifting the YOK-I datasets ahead of the dTSI data appropriately. Using a window of 450 years over the data, the evolution of lagged correlation was obtained which helped demarcate distinct time periods based on the behavior of the lagged correlation values over time (Fig. 8B).

The analysis reveals the presence of statistically significant correlations with a persistent lag between 30 and 50 years of  $DCF_{YOK-I}$  with respect to dTSI during the period 900-1250 C.E. However, for the period after  $\sim 1250$  C.E., we fail to

detect similar statistically significant correlations. The same analysis was also performed on  $\delta^{13}\text{C}$ , yielding very similar results as  $\text{DCF}_{\text{YOK-I}}$  (although the lag extends between 10-50 years in this case) (Fig. 8B). These observations strongly suggest that hydrologic change at Yok Balum Cave occurred several decades after shifts in atmospheric  $^{14}\text{C}$  content, induced by solar irradiance, and were not a direct reflection of contemporaneous atmospheric  $^{14}\text{C}$ . Rainfall at Yok Balum Cave is largely controlled by the seasonal migration of the ITCZ, and due to the cave's location at the present-day northern boundary of the annual ITCZ range, stalagmites from the site are very sensitive to subtle southward ITCZ migration (Ridley et al., 2015a). Because the ITCZ tracks the Earth's thermal equator, it migrates in response to hemispheric and global temperature shifts (Schneider et al., 2014), controlled by the strength of the Sun, which also modulates atmospheric  $^{14}\text{C}$  content. Two possible processes could induce a lagged response to the atmospheric records in  $\text{DCF}_{\text{YOK-I}}$ :

- i) The stalagmite DCF is influenced by a large pool of 'old' organic carbon derived from the soil or deep vadose zone, or
- ii) The lag is an actual reflection of a delayed response of rainfall patterns at Yok Balum Cave to solar forcing on climate.

The presence of large amounts of old carbon in the karst system is unlikely, because the model results (based on the YOK-I bomb spike) suggest otherwise. In addition, the fact that the lagged response to solar forcing is detectable in both DCF and  $\delta^{13}\text{C}$  (and U/Ca) suggests that there is another factor influencing both proxies. Numerous studies have found decadal-scale lags (on the order of 10-40 years) in the response of rainfall patterns to solar forcing (Kobashi et al., 2015; Moffa-Sanchez et al., 2014; Shindell et al., 2001; Swingedouw et al., 2011; Waple

et al., 2002). It is possible that a similar delayed response of Mesoamerican rainfall to solar forcing results in the lag observed in  $DCF_{YOK-I}$ , especially prior to 1250 C.E. Although the precise nature of the observed lag is unclear, solar-induced changes in the amount of freshwater and/or sea ice delivered to the North Atlantic basin and subsequent feedbacks in oceanic and atmospheric dynamics (e.g., in the state of the North Atlantic Oscillation) may play a role (Kobashi et al., 2015; Swingedouw et al., 2011; Waple et al., 2002). A possible solar influence on drought occurrence in the Yucatan has previously been suggested by Hodell et al. (2001). The apparent weakening of the solar influence on the YOK-I record after 1250 C.E. suggests a shift in the mechanism responsible for the observed lag between solar activity and rainfall at Yok Balum Cave. Although the causes for the lagged response between  $DCF_{YOK-I}$  and solar activity remain unclear, we note that the breaking down of the lagged proxy-Sun relationship (potentially a complete decoupling between rainfall and solar activity) is broadly synchronous with the beginning of the Little Ice Age (LIA), a period of extensive cooling in the Northern Hemisphere (Mann et al., 2009). This could therefore reflect a decreased influence of solar activity on hydroclimate (at least in Mesoamerica) during the LIA, and emergence of a different dominant forcing on ITCZ position (e.g., volcanism, Miller et al., 2012). However, other climate reconstructions and more extensive research are required to verify this interpretation.

## **6. Conclusions**

We present a comprehensive study of carbon cycling and the controls on stalagmite DCF at the tropical Yok Balum Cave, southern Belize. Subdecadal-scale



DCF,  $\delta^{13}\text{C}$ , and U/Ca records from stalagmite YOK-I covering the last 1500 years, combined with bulk SOC, WEOC, and modeling analysis of  $^{14}\text{C}$ , reveal the sources of carbon incorporated in stalagmite YOK-I, and on the factors and processes that give rise to variations in DCF:

- Overall, the largest contribution to total  $\text{DCF}_{\text{YOK-I}}$  is carbonate bedrock dissolution in the karst, significantly modulated by hydrological conditions. Contributions of SOM to the total  $\text{DCF}_{\text{YOK-I}}$  are relatively small, due to the fast SOM turnover and low carbon storage potential of the soil. Dynamic ventilation of the cave system and seasonal aridity in the region results in strong kinetic fractionation effects and PAP acting on  $\delta^{13}\text{C}$  and U/Ca. We acknowledge, however, that our approach of using constant vegetation and SOM parameters in the model might bear some weaknesses and should be refined by future studies.
- We find a strong relationship between DCF,  $\delta^{13}\text{C}$ , and U/Ca, suggesting a common forcing factor on all three proxies (i.e., hydroclimate conditions above the cave). These results highlight the potential usefulness of  $\delta^{13}\text{C}$  and trace element ratios to track changes in stalagmite DCF, and could help detecting past shifts in DCF when no independent age control is available (e.g., for periods beyond the tree-ring based interval of the atmospheric  $^{14}\text{C}$  calibration curve) or for stalagmite  $^{14}\text{C}$  dating purposes.
- Comparison of the high-resolution  $\text{DCF}_{\text{YOK-I}}$  and  $\delta^{13}\text{C}$  records to IntCal13 and solar records shows compelling similarity with a variable lag (10-50 years) in the response of YOK-I to the solar forcing. We suggest that rainfall above the site was driven by solar forcing but with a lagged response, and raise the possibility that solar forcing of ITCZ position

varies temporally, and becomes much less prominent after the transition  
into the LIA.

#### **Acknowledgements:**

The authors gratefully acknowledge generous help from the LIP staff members,  
especially L. Wacker during sample preparation and measurement. N. Haghypour  
is thanked for help during sample preparation and accelerator mass  
spectrometry analysis. A.E. Thompson from the Uxbenká Archaeological Project  
is thanked for providing the map of the study site. C. Ottley is thanked for the  
ICP-MS aspects of the research. F. Hagedorn from the Swiss Federal Institute for  
Forest, Snow, and Landscape research (WSL) is thanked for advice regarding the  
WEOC methodology. This research was supported by the European Research  
Council grant 240167 to JULB.

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#### **Tables:**

Table 1: Results of the proxy study on stalagmite YOK-I. DCF was calculated

using the formula:  $DCF = 1 - \left( \frac{a^{14}C_{stal.init.}}{a^{14}C_{atm.init.}} \right)$

where  $a^{14}C_{stal.init.}$  and  $a^{14}C_{atm.init.}$  represent stalagmite and atmosphere  $^{14}C$  activity (respectively) at the time of carbonate deposition.

Table 2: Analysis of soil samples. A profile consisting of 13 samples was collected above Yok Balum Cave, and both bulk SOC and WEOC  $^{14}C$  were measured.

#### **Figures:**

Fig. 1. Maps of Yok Balum cave, southern Belize: A - Cave map showing the location of stalagmite YOK-I (red dot), and the approximate position of the soil profile collected above the cave (yellow circle) (Map A is adapted from a map by Tom Miller). B - Topographic map of the study site, with indication of the location of Yok Balum cave. C - Overview map of Central America and the general setting of the cave. (Maps in B and C are by A.E. Thompson, courtesy of the Uxbenká Archaeological Project).

Fig. 2: A - Photoscan of the section of stalagmite YOK-I that was analysed for  $^{14}\text{C}$  together with results from  $^{14}\text{C}$  analysis (U/Th ages are shown for comparison). A pronounced bomb spike appears at the top of the stalagmite. B - Bomb spike recorded in stalagmite YOK-I (red dots), compared to the atmospheric record from the northern Hemisphere zone 2 (Hua et al. (2013), black line), and to the bomb spike recorded in stalagmite YOK-G from the same cave (grey dots, Ridley et al. (2015a)). Signal damping due to the age spectrum of SOM results in the lower amplitude and slightly delayed response of stalagmites YOK-I and YOK-G with respect to the atmosphere.

Fig. 3: Results of the analysis of geochemical proxies in stalagmite YOK-I: A - DCF calculated from  $^{14}\text{C}$  measurements (purple line, including  $1\sigma$  errors); B - U/Ca in ppm/ppm  $\times 1000$  (green line); C -  $\delta^{13}\text{C}$  measured on the same aliquots as used for  $^{14}\text{C}$  analysis (dark red diamonds) show that no sampling bias occurred with respect to the original high-resolution  $\delta^{13}\text{C}$  time series (light red line); D -  $\delta^{18}\text{O}$  from the original high-resolution time series (both high resolution stable isotope records were previously published in Kennett et al., 2012).

Fig. 4: Results from the analysis of the soil profile collected above Yok Balum cave. Amount of carbon present in the samples was determined twice, showing very reproducible results (grey and black dots).  $F^{14}\text{C}$  shows regularly decreasing values through the bulk SOC profile with bomb carbon imprint in the top 10 cm (red dots), and a slower decrease in the WEOC (blue dots).

Fig. 5: Results of the modeling procedure on stalagmite YOK-I: A - The best fit of the model with the bomb spike data (black symbols) is shown by the red dashed line, the atmospheric bomb spike is shown for comparison in blue. B - The calculated soil air  $^{14}\text{C}$  activity, after applying the SOM spectrum derived from the bomb spike on the entire time series, is shown in green. Fractionation between gaseous  $\text{CO}_2$  and DIC results in slight enrichment (orange line). The atmospheric activity is shown in black for comparison. C - Results of the deconvolution of DCF: the black line shows the total DCF as measured on the stalagmite. The DCF contribution from vegetation/SOM is shown by the green line, and in-cave fractionation effects result in the orange line. DCF derived from host rock dissolution is shown in purple.

Fig. 6: Evolution of  $\delta^{13}\text{C}$  in the Yok Balum karst system, determined by modeling. A -  $\delta^{13}\text{C}$  of the drip water (blue line), calculated using the measured total DCF, indicating the degree of open vs. closed system and consequently soil  $\text{CO}_2$  exchange with the aqueous solution in the karst. B - U/Ca (green line) is modulated by PAP, and shows remarkable similarity with stalagmite  $\delta^{13}\text{C}$ . C -  $\Delta\delta^{13}\text{C}$  (black line) is calculated as the difference between  $\delta^{13}\text{C}$  of the drip water and the stalagmite, and reflects the amount of kinetic fractionation affecting the sample (as described in Griffiths et al. (2012)). D -  $\delta^{13}\text{C}$  in YOK-I (red line), underlain by the high resolution profile presented in Kennett et al. (2012).

Fig. 7: Relationship between  $\delta^{13}\text{C}$ , U/Ca and DCF in stalagmite YOK-I: A - Relationship between  $\delta^{13}\text{C}$  and U/Ca. A significant linear correlation (black line,  $r = -0.83$ ,  $p < 0.001$ ; 95% confidence interval as grey dashed line) exists between

$\delta^{13}\text{C}$  and U/Ca ratios. DCF values are color-coded. B - Scatterplots showing the relationship between DCF and  $\delta^{13}\text{C}$  (upper), and DCF and U/Ca (lower), with associated correlation coefficients. All proxies are influenced by karst infiltration:  $\delta^{13}\text{C}$  reflects the amount of PCP/PAP and kinetic fractionation in the cave, whereas U/Ca is influenced by PAP. DCF responds to the degree of open-vs-closed system conditions in the karst, modulated by changes in effective infiltration.

Fig. 8: External forcing on YOK-I carbon isotopic records: A - Comparison of YOK-I DCF and  $\delta^{13}\text{C}$  to total solar irradiance (dTSI) calculated from  $^{10}\text{Be}$  (blue curve) (Steinhilber et al., 2009) and atmospheric  $\Delta^{14}\text{C}$  from IntCal13 (grey curve) (Reimer et al., 2013). Dashed lines indicate features present in all records suggesting solar forcing with a variable lag on precipitation at Yok Balum Cave. U/Th ages for stalagmite YOK-I are shown to highlight the excellent age control of the record. B - Lag-correlation plots quantifying the lag between DCF and  $\delta^{13}\text{C}$  at Yok Balum Cave and dTSI (Steinhilber et al., 2009). The test was done with 5000 randomized surrogates for each lag. Dashed lines indicate significant values (two-sided at 0.05 significance level with Bonferroni correction). A band of high correlations lagging the solar forcing by  $\sim 30$ -50 years for DCF, and  $\sim 10$ -50 for  $\delta^{13}\text{C}$  is visible between  $\sim 900$ -1250 C.E., but the relationship breaks down in the younger part of the record (1300-1700 C.E.). Note that  $\delta^{13}\text{C}$  is plotted on an inverse colorbar compared to DCF since the two proxies have the opposite response to hydrological changes (wetter: DCF increases,  $\delta^{13}\text{C}$  decreases).

Suppl. Fig. 1: Conceptual diagram of carbon cycle processes occurring in a karst system and the associated response in the hydrological proxies ( $^{14}\text{C}$ ,  $\delta^{13}\text{C}$  and  $\text{U/Ca}$ ) used in this study.

Suppl. Fig. 2: Schematic of the modeling process (as in Griffiths et al., 2012) as applied to the dataset from stalagmite YOK-I. The model is composed of two parts: in a first step, the stalagmite bomb spike is used to calculate the best fitting SOM spectrum, by using a Monte Carlo optimization process. In the second part, the SOM spectrum is applied to the remaining stalagmite dataset and the contributions to DCF from vegetation, in-cave fractionation and host rock dissolution can be separated and quantified.